

# Synthesis of 1,2,3,7,8,9-Hexahydrodipyrrens and Secocorrins:<sup>†</sup> Important Precursors for the Construction of Corrins

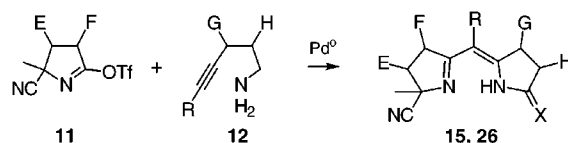
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## ABSTRACT



Hexahydrodipyrrens **15** ( $X = H_2$ ) have been prepared by two routes: Pd(0)-initiated coupling/cyclization of triflates **11** and alkyne amines **12** and reduction of semicorrins **26** ( $X = O$ ).

Tetrapyrroles of the chlorin, bacteriochlorin, and corrin oxidation level serve important functions in nature, and they continue to attract considerable attention.<sup>1</sup> Part of this interest derives from the isolation of novel hydroporphyrins from both marine and microorganisms, some of which have intriguing biological activity.<sup>1,2</sup> In addition, chlorins and bacteriochlorins have potential utility in photodynamic therapy (PDT).<sup>3</sup> The most complex members of this class are the corrins, which can contain up to 10 stereogenic centers on the macrocycle.

In 1977, Eschenmoser et al. published a synthesis of cobyrinic acid (**6**) that made ingenious use of the recently discovered Woodward–Hoffmann rules.<sup>4a</sup> Thus, rings A–D

(**1–4**) were coupled employing sulfide contraction methodology to afford the secocorrin derivative **5**, which was maintained in the all-*Z* geometry by complexation with Cd (Figure 1). Secocorrin **5** then underwent a symmetry-allowed, antarafacial 1,16-hydrogen transfer upon exposure to visible light, and the resultant intermediate, upon cycloisomerization, afforded the corrin skeleton of **6** with >95% stereoselectivity. This strategy represents a dramatic improvement over traditional approaches involving coupling of AD + BC ring fragments,<sup>4b</sup> and its implementation was an extraordinary achievement. The versatility of this method is limited only by the availability of complex intermediates of type **1–4** and the fact that the sulfide contraction procedure is not well suited for introducing meso substituents.<sup>5</sup> As a consequence, peripheral substituents are generally introduced after macrocycle formation, in what is typically a low-yielding process (cf. the C<sub>5</sub> and C<sub>15</sub> Me substituents in **6**).<sup>5b</sup> In addition, the lower oxidation state of ring D can present special challenges.

<sup>†</sup> Nomenclature of Tetrapyrroles (Recommendations 1986). *Pure Appl. Chem.* **1987**, 59, 779.

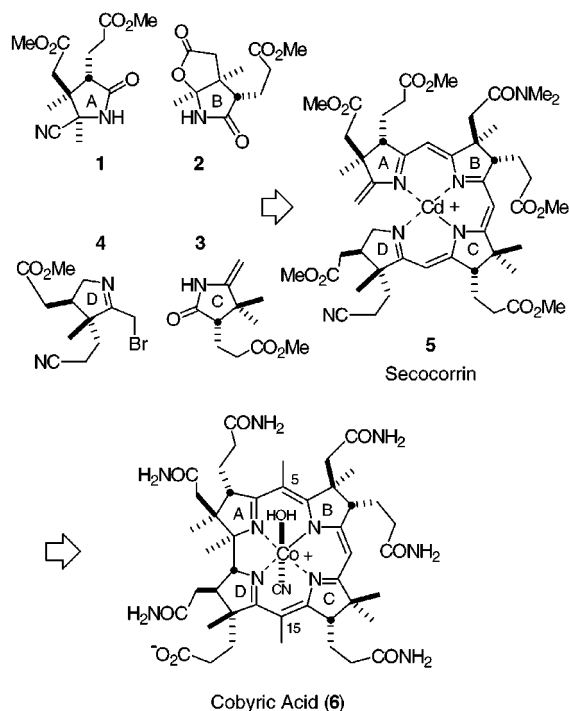
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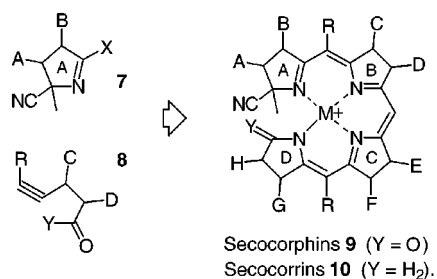
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**Figure 1.** Eschenmoser's cobyrinic acid synthesis.

Recently, we described an iterative synthesis of secocorphanes **9** that took advantage of the ready availability of imidoyl derivatives **7** ( $X = \text{Tf}, \text{Cl}$ ) and alkyne acids/amides **8** ( $R = \text{H}, \text{Me}$ ;  $Y = \text{OH}, \text{NH}_2$ ) (Figure 2).<sup>6a,b</sup> A key step in

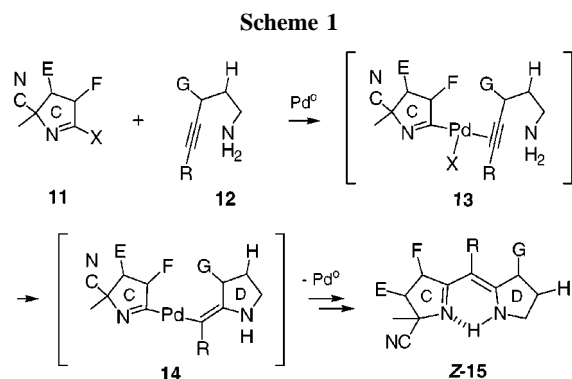


**Figure 2.** Iterative synthesis of secocorphanes **9**.

this synthesis was the  $\text{Pd}(0)$ -initiated coupling/cyclization of **7** and **8** to afford the corresponding ring AB- and CD-semicorrins. Linking of these fragments employing Eschenmoser's sulfide contraction procedure then gave **9** in good to very good yields ( $A, C, E, G = \text{Me}_2$ ;  $B, D, F, H = \text{H}$ ).<sup>6</sup> The presence or absence of meso substituents at  $C_5$  and  $C_{15}$  was dictated solely by the choice of starting alkyne acid/amide **8** (i.e.,  $R = \text{H}$  or  $\text{Me}$ ).

In principle, reduction of ring D in **9** could provide secocorrins **10** having both the proper oxidation state and substitution pattern for cyclization to corrins (cf. also **5** in

Figure 1). In practice, however, this transformation is difficult to achieve.<sup>7</sup> As alternate routes to **10**, we have explored two strategies for preparing ring CD-semicorrins where the terminal ring is in the pyrroline oxidation state. Our first approach follows the precedent set in the coupling of imidoyl derivatives **7** and alkyne acids **8** (cf. Figure 2)<sup>6</sup> and takes advantage of the nucleophilic character of alkyne amines **12** in  $\text{Pd}(0)$ -initiated coupling/cyclization reactions (Scheme 1).<sup>8</sup>



In this route, oxidative addition of  $\text{Pd}(0)$  to imidoyl derivatives **11** ( $X = \text{Tf}, \text{Cl}$ ) would be followed by  $\pi$ -complexation and then nucleophilic capture to generate vinyl- $\text{Pd}(\text{II})$  species of type **14** ( $\text{Pd}$  ligands are not shown for the sake of clarity). Finally, reductive elimination would afford ring CD amidine derivatives **15** ( $R = \text{H}, \text{alkyl}$ ), presumably as the *E* isomers (retention of configuration).<sup>9</sup> However, we expected that (*E*)-**15** would undergo rapid equilibration to the more stable (*Z*)-**15**, which contains an internal hydrogen bond. As previously noted,<sup>6b</sup> the thermodynamic driving force for expulsion of  $\text{Pd}(0)$  in **14** should overcome any steric crowding in the transition state leading to **15**.

A potential advantage of this approach is the relative mildness of the steps employed. In addition, we expected that chiral alkyne amines of type **12** could be readily derived by reduction of the corresponding amides, which we have previously prepared using a modified Nicholas-Schreiber reaction.<sup>10</sup> In model studies, this reduction was achieved in excellent yield by treatment of alkyne amides **16** with LAH/ $\text{Et}_2\text{O}$  (Scheme 2). Alkyne amines **17a–e** were chosen to evaluate the effects of steric hindrance and alkyne substitution pattern on reactivity.

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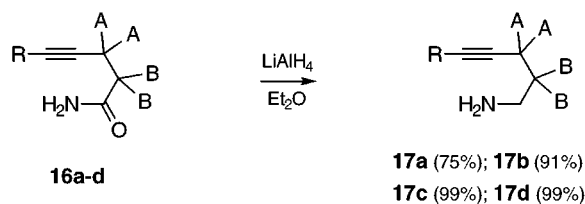
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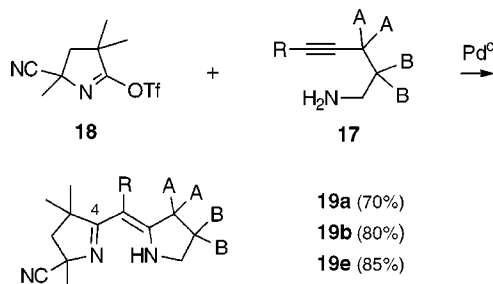
Scheme 2



a: A = H, B, R = Me. b: A, R = H; B = Me.  
 c: A = Me; B, R = H. d: A, R = Me; B = H. e: A, B, R = H.

Our initial experiments were conducted with the iminoyl triflate **18**, whose synthesis we have previously described (Scheme 3).<sup>6a</sup> In this compound, the geminal methyl groups

Scheme 3

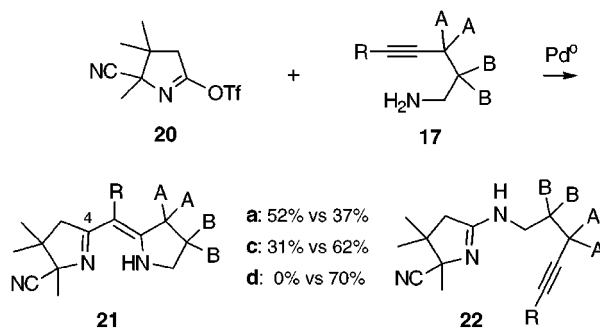


a: A = H; B, R = Me. b: A, R = H; B = Me. e: A, B, R = H.

at C<sub>3</sub> provide a useful probe of steric hindrance at C<sub>4</sub> (semicorrin numbering). We were pleased to find that Pd(0)-initiated coupling/cyclization of **18** with alkyne amines **17** occurred cleanly and afforded 70–85% yields of the corresponding (*H*<sub>6</sub>)-dipyrrins **19**. For example, **18** combined at room temperature with terminal alkynes **17b** and **17e** (R = H), using the reagent system Pd(Ph<sub>3</sub>P)<sub>4</sub>/NEt<sub>3</sub>/THF (80% and 85% yields, respectively). Similarly, reaction of alkyne amine **17a** (R = Me) with **18** gave a 70% yield of the meso-substituted (*H*<sub>6</sub>)-dipyrrin **19a**. This last conversion required warming to 80 °C (MeCN), and was accelerated by added BnN(Et)<sub>3</sub>Cl (rate enhancements by ammonium salts have been observed previously).<sup>6b,9</sup> Also, the most effective catalyst was the Pd<sub>2</sub>dba<sub>3</sub>/TFP system described by Farina.<sup>11</sup> Importantly, the effect of steric hindrance at C<sub>4</sub> on these Pd(0)-initiated reactions appeared to be minimal.

The results obtained with the isomeric iminoyl triflate **20** were less straightforward (Scheme 4). In **20**, the geminal methyl groups at C<sub>2</sub> are well removed from the reacting center (C<sub>4</sub>) and, therefore, provide little steric shielding. Consequently, direct nucleophilic displacement at C<sub>4</sub> competes effectively with the Pd(0)-initiated process. In the best case, coupling of **20** with the alkyne amine **17a** gave a 52%

Scheme 4



a: A = H; B, R = Me. c: A = Me; B, R = H. d: A, R = Me; B = H.

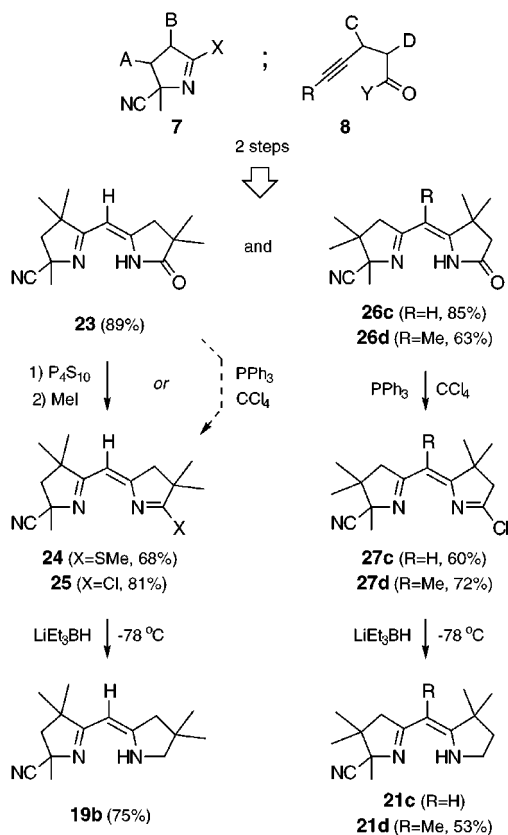
yield of the meso-substituted (*H*<sub>6</sub>)-dipyrrin **21a** (A = H; B, R = Me), accompanied by 37% of amidine **22a**, the product of amine displacement. In analogous fashion, the more sterically hindered alkyne **17c** afforded only 31% of the desired (*H*<sub>6</sub>)-dipyrrin **21c** (A = Me; B, R = H) and 62% of amidine **22c**. Finally, alkyne amine **17b**, which contains the most shielded triple bond (A, R = Me) and the least hindered amine (B = H), gave amidine **22d** as the exclusive product (70% yield). In summary, steric crowding at C<sub>4</sub> has little or no effect on the desired Pd(0)-initiated coupling/cyclization. In fact, it can be beneficial in retarding undesired side reactions. However, shielding of the alkyne bond by both terminal and adjacent methyl substituents has a pronounced effect on selectivity and generally favors direct attack by the amine.

Nucleophilic displacements of the type described in Scheme 4 do not occur with alkyne acids/amides **8**,<sup>6</sup> which are excellent substrates for Pd(0)-initiated coupling and cyclization (Scheme 5). For example, semicorrins **23** and **26c,d** were prepared in two steps, and in high yield, from the appropriate imidoyl derivatives **7** and alkyne acid derivatives **8**.<sup>6</sup> In principle, reduction of semicorrin **23** could provide an alternate route to the (*H*<sub>6</sub>)-dipyrrin **19b**, which we had previously synthesized using our coupling/cyclization methodology (cf. Scheme 3). If successful, this approach might then be applied to the synthesis of less accessible (*H*<sub>6</sub>)-dipyrrins of type **21**.

Surprisingly, there appear to be no reports describing the reduction of semicorrins of type **23** to 1,2,3,7,8,9-hexahydridipyrrins. In any event, this transformation proved to be very challenging. For example, all attempts at the direct reduction of **23** utilizing reagents such as Et<sub>3</sub>OBF<sub>4</sub>/NaBH<sub>4</sub>, BH<sub>3</sub>·Me<sub>2</sub>S/BF<sub>3</sub>·Et<sub>2</sub>O, LAH, and LiEt<sub>3</sub>BH were unsuccessful. In every case, we observed either no reaction or, under more vigorous conditions, severe decomposition. None of the desired **19b** could be detected by TLC analysis.

We also explored the possibility of reducing the thioimino-ester **24**, which was readily derived from **23** by initial thiolactam formation (P<sub>4</sub>S<sub>10</sub>; 78%), followed by methylation (MeI/DBU; 87%). In most cases, **24** gave intractable mixtures

Scheme 5

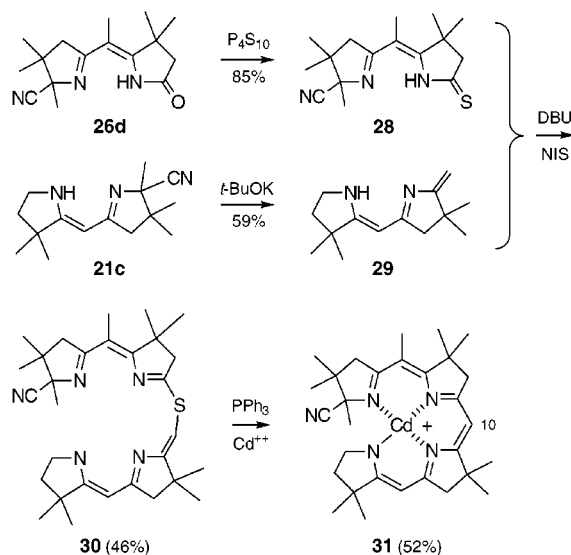


upon attempted reduction.<sup>12a</sup> With  $\text{LiBH}_4$  in THF, however, we were able to detect trace amounts of **19b** in the crude reaction mixtures. Encouraged by this result, we prepared the more reactive imidoyl chloride **25** ( $\text{PPh}_3/\text{CCl}_4$ ; 81%) and tested the same range of reducing agents.<sup>12a</sup> Finally, we were pleased to find that the desired transformation could be effected in 75% yield utilizing Super-Hydride ( $\text{LiEt}_3\text{BH}$ ) in THF at  $-78^\circ\text{C}$ . In identical fashion, **26c,d** gave the corresponding imidoyl chlorides **27c,d** and **27d** afforded the ( $H_6$ )-dipyrin **21d** (53%).

Having achieved practical syntheses of both semicorrins **26** and ( $H_6$ )-dipyrins **21**, we employed Eschenmoser's

(12) (a) Reagents:  $\text{NaBH}_4$ , Raney Ni,  $\text{NaBH}_4/\text{BF}_3\cdot\text{Et}_2\text{O}$ ,  $\text{NaBH}_4/\text{ZnCl}_2$ ,  $\text{Et}_3\text{SiH}/\text{BF}_3\cdot\text{Et}_2\text{O}/\text{NaBH}_4$ ,  $\text{LiEt}_3\text{BH}$ , and  $\text{LiBH}_4$ . (b) Satisfactory analytical and spectral data were obtained for all new compounds.

Scheme 6



methodology to synthesize the meso-substituted secocorrin **31** (Scheme 6).<sup>4b</sup> This required the initial preparation of thiolactam **28** and dipyrin derivative **29**, both of which were readily obtained following literature precedent (cf. Supporting Information).<sup>4b,6b</sup> Oxidative coupling of **28** with **29** was then accomplished in 46% yield using NIS in the presence of DBU.<sup>2,4b</sup> Finally, treatment of sulfide **30** with  $\text{PPh}_3$  and  $\text{CdCl}_2$  led directly to the secocorrin **31** (52%), which is in the correct oxidation state for photochemical cycloisomerization.<sup>4a,12b</sup>

Following a similar approach to that described above, we believe that this methodology might be extended to the synthesis of cobyric acid (**6**) and other complex, naturally occurring tetrapyrroles. These studies are in progress.

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**Supporting Information Available:** Copies of  $^1\text{H}$  and/or  $^{13}\text{C}$  NMR spectra and experimental procedures for compounds **17**, **19**, **20–22**, **24**, **30**, and **31**; UV-vis spectra for compound **31**. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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